

Development of active catalyst for direct conversion of biomass cellulose to C₃ and C₄ hydrocarbons at low temperature

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Abstract: Direct catalytic conversion of cellulose to C₃ and C₄ hydrocarbons at low temperature (443 K) in the presence of Pt/zeolite catalysts and water was investigated. Activity tests revealed that Pt supported on H⁺-form ultra stable Y type (H-USY) zeolite catalyst (Pt/H-USY) enabled direct conversion of cellulose into C₃ and C₄ hydrocarbons without hydrogen and other expensive reagents. This production trend is attributable to the bifunctional catalysis of supported Pt and acid site in H-USY zeolite. Results revealed that the pretreatment of catalyst was important to improve C₃ and C₄ olefin selectively. Air-oxidized Pt/H-USY catalyst showed higher C₃ and C₄ olefin selectivity than the catalyst reduced with hydrogen.

Keywords: Cellulose, C₃ and C₄ olefin, Hydrothermal reaction

1. Introduction

Renewable resources are attracting to contribute sustainable society. In particular, biomass cellulose offers numerous benefits as a ubiquitous feedstock. Recently, although C₃ and C₄ hydrocarbons are important raw materials used in petrochemistry, there is concerned about the shortage of these hydrocarbons supply in the future due to the stop of naphtha cracker (main light olefin production process). Therefore, development of new C₃ and C₄ hydrocarbons production process is anticipated. We focused on direct catalytic reaction of cellulose to C₃ and C₄ hydrocarbons. Recent reports described that a one-pot direct catalytic conversion of cellulose to C₃ and C₄ hydrocarbons over Pt/zeolite catalysts at 443 K under hydrothermal conditions without external hydrogen.¹⁻³⁾ The activity and selectivity in the reaction were affected by the zeolite structure and the Pt supported state. In the present work, to improve the olefin selectivity, effect of the pretreatment of catalyst on C₃ and C₄ olefin selectivity was investigated.²⁾

2. Experimental

Pt/H-USY catalyst was prepared using an ion exchange method. 1wt% Pt was supported on H-USY(14) (Tosoh Corp.) zeolite. The number in parenthesis representing the SiO₂/Al₂O₃ ratio. Catalyst was calcined at 803 K for 8 h. Catalyst was pretreated in a fix-bed continuous-flow glass reactor with 10% H₂ flow or air flow (100 ml/min) at 773 K for 1 h. Before the reaction, cellulose powder (MP Biomedicals) was pretreated using a ball mill (Pulverisette 6 classic line; Fritsch GmbH) to reduce its crystallinity.

Catalytic activity tests were performed in a batch-wise autoclave (30 ml, SUS316) with the following reaction conditions: 20 ml water, 0.25 g (8330 C- μ mol) ball-milled cellulose, 0.25 g catalyst, 443 K temperature, 600 rpm stirring rate, 1-24 h reaction time. Catalytic reactions were conducted under Ar atmosphere. The reaction starting time was recorded when the reaction temperature reached the preset temperature. After the reaction, the reactor was cooled with water for 30 min. Gaseous products were collected and analyzed using GC-FID. Liquiform products were analyzed using TOC and HPLC.

3. Results and discussion

To improve C₃ and C₄ olefin selectivity, the effect of catalyst pretreatment on the catalytic activity and selectivity were investigated. Figure 1 shows C₃ and C₄ hydrocarbon selectivities over the air-oxidized and H₂-reduced Pt/H-USY zeolite catalysts. From Figure 1, air-oxidized Pt/H-USY catalyst showed higher olefin

selectivity than that over the H₂-reduced Pt/H-USY catalyst. On the other hand, cellulose conversion and gaseous product amounts were almost identical in both pretreatment conditions. Results indicated that the oxidation pretreatment improved C₃ and C₄ olefin selectivities.

To clarify the difference of selectivity, TEM and EXAFS measurements were conducted. From TEM images, no significant difference in average Pt particle size was observed. Figure 2 shows Pt L₃-edge EXAFS spectra after Fourier transformation for air-oxidized and H₂-reduced Pt/H-USY(14) catalysts. Only Pt-Pt bond was observed over H₂-reduced Pt/H-USY(14) catalyst. On the other hand, both Pt-O bond and Pt-Pt bond were observed for air-oxidized Pt/H-USY(14) catalyst.²⁾ It was indicated the supported Pt was partially oxidized. Therefore, partially oxidized Pt species (PtO_x) might be effective for C₃ and C₄ olefin production.

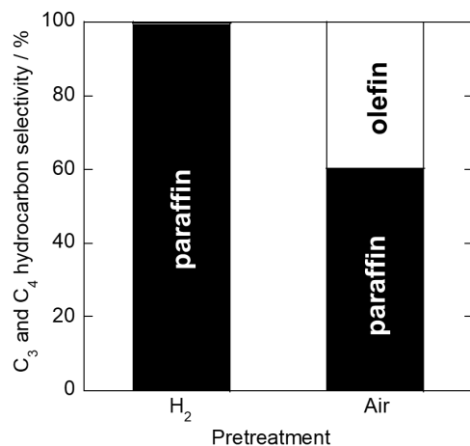


Figure 1. Effect of catalyst pretreatment on C₃ and C₄ hydrocarbons selectivity in the cellulose conversion over Pt/H-USY(14) catalyst at 443 K for 3 h.

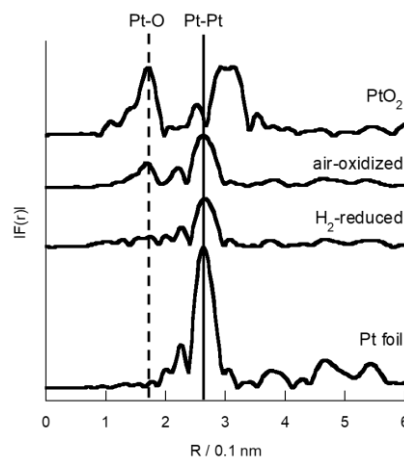


Figure 2. EXAFS spectra for Pt L₃-edge after Fourier transformation over Pt/H-USY(14) catalysts after pretreatment under H₂ or air flow at 773K for 1 h.²⁾

Next, we investigated the effect of reaction time on the olefin selectivity over air-oxidized Pt/H-USY(14) catalyst. Catalytic conversion of cellulose over Pt/H-USY(14) catalyst was conducted at 443 K for 1-24 h. Cellulose conversion and C₃ and C₄ hydrocarbon yield increased with increased reaction time. On the other hand, C₃ and C₄ olefin selectivities decreased gradually with increased reaction time up to 7 h, and then drastically decreased after further increase of reaction time. These results suggested that hydrogenation of olefin to paraffin proceeded with increased reaction time.

Then, to clarify the reaction intermediates, catalytic conversion of various substrates (cellobiose, glucose, fructose and xylose) were conducted over the Pt/H-USY(14) catalyst at 443 K for 3 h. In cases for cellobiose, glucose and fructose conversions, C₄ hydrocarbons were formed with high selectivity. In the case of xylose conversion, C₃ hydrocarbons were formed with high selectivity. These results indicate that C₆ sugars are reaction intermediates for C₄ hydrocarbon formations and C₅ sugars are reaction intermediates for C₃ hydrocarbon formations.

4. Conclusions

Air-oxidized Pt/H-USY catalyst showed higher olefin selectivity than that on H₂-reduced catalyst. Catalyst pretreatment condition influenced the oxidation state of the supported Pt and thereby changed olefin selectivity. C₃ and C₄ hydrocarbon yields increased with increased reaction time while C₃ and C₄ olefin selectivities decreased. C₆ sugars are reaction intermediates for C₄ hydrocarbons and C₅ sugars are reaction intermediates for C₃ hydrocarbons from cellulose over Pt/H-USY(14) catalyst.

References

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